

## A Novel Construction of 2-Benzazepine Scaffold Based on TiCl<sub>4</sub>-Mediated Tandem *Mannich* Reaction – Aromatic Electrophilic Substitution

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A novel construction of 2-benzazepine derivatives based on TiCl<sub>4</sub>-mediated tandem *Mannich* reaction of electron-rich benzyl iminium ions with alkenyl ethers and *Friedel–Crafts*-type alkylation is described. The protocol is amenable to provide the tricyclic furo[3,2-*d*][2]benzazepine and pyrano[3,2-*d*][2]benzazepine derivatives, respectively, with 2,3-dihydrofuran or 3,4-dihydro-2*H*-pyran as the substrates.

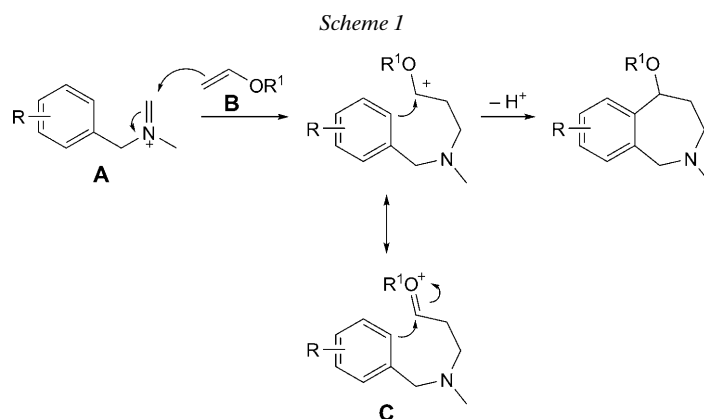
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**Introduction.** – The 2-benzazepine motif is a core structure in a number of pharmacologically important compounds [1]. Several members of this class have been found to exhibit hypotensive, anticonvulsant, analgesic, and antiarrhythmic activities, and also have been proved to be useful as antagonists of muscarinic (M<sub>3</sub>) receptors and for the treatment of mental disorders and hypoxia [2][3]. Many naturally occurring molecules also contain a 2-benzazepine skeleton. For example, the alkaloids of the *Amaryllidaceae* group including galanthamine, lycoramine, and narwedine are tetrahydro-2-benzazepine derivatives [1][4][5]. In addition, many simpler synthetic analogues have been disclosed to possess a broad spectrum of biological activities [6]. Accordingly, considerable attention has been devoted to the development of convenient and efficient synthesis of families of this privileged heterocyclic ring system [1].

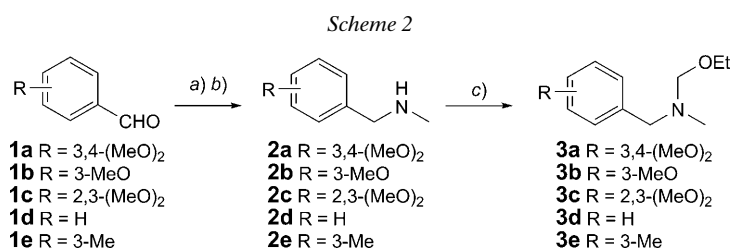
The most widely used methodologies involve *Pictet–Spengler* cyclization [7], the *Schmidt* reaction on tetralones [8], cyclization of benzyl amine and cinnamyl chloride [9], a ring-closing metathesis strategy [10], and transformation of the *Baylis–Hillman* adducts via tandem *Ritter* and *Houben–Hoesch* reactions [11]. A recent article by *Nagumo et al.* reported a novel construction of 2-benzazepines based on TMSOTf-promoted 7-*endo* *Friedel–Crafts*-type reaction of vinyloxiranes [12]. However, these approaches require multistep sequences or expensive reagents.

Our strategy for developing a novel synthesis of 2-benzazepines is outlined in *Scheme 1*. We envisioned that the initial reaction of benzylic iminium ions **A** with vinyl ethers **B** may give rise to oxonium ions **C** with formation of a C–C bond. The intermediates **C** will be capable of concurrent *Friedel–Crafts*-type alkylation to give the 7-*endo-trig* cyclization products, *i.e.*, 2-benzazepines, by formation of a new C–C bond. Remarkably, there is no precedent of this tandem synthesis of 2-benzazepines.

**Results and Discussions.** – To test our hypothesis, we firstly synthesized from **1a** via **2a** the aminoacetal **3a** as the precursor of the required benzyl iminium ion **4a**. Thus,



benzaldehyde **1a** was subjected to reductive amination with MeNH<sub>2</sub> to afford *N*-benzylmethylamine **2a**, which was readily converted to **3a** by reaction with paraformaldehyde and anhydrous EtOH under basic conditions [13] (Scheme 2).

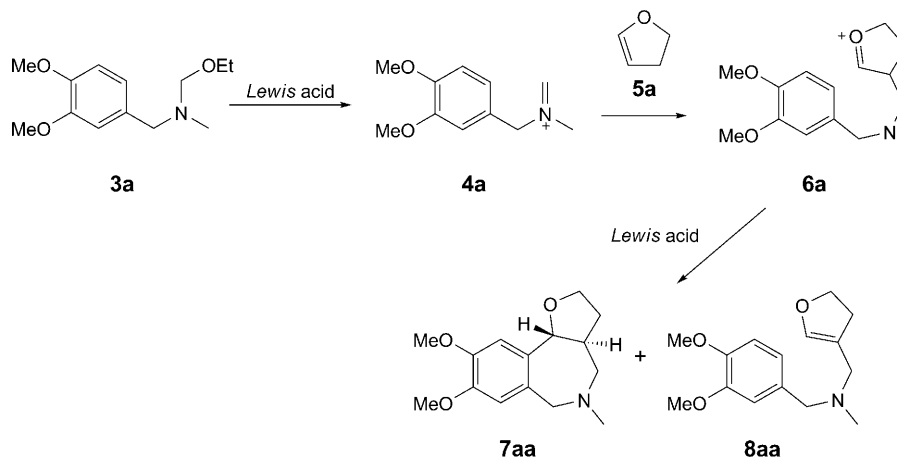


a) MeNH<sub>2</sub>, C<sub>6</sub>H<sub>6</sub>, reflux, 2 h; b) NaBH<sub>4</sub>, EtOH, 3 h (80–85%); c) (CH<sub>2</sub>O)<sub>n</sub>, EtOH, K<sub>2</sub>CO<sub>3</sub>, 1–2 d (95%).

The aminoacetal **3a** and 2,3-dihydrofuran **5a** were selected as the substrates to screen the appropriate *Lewis* acid used for the generation of the benzyl iminium ion **4a** (Scheme 3 and Table 1). If **4a** can be formed *in situ*, it should undergo an electrophilic addition with the pre-added **5a** to give the oxonium ion **6a**. Electrophilic aromatic substitution on the *ortho*-position of the Ph ring would afford the tricyclic furobenzazepine **7aa**. Examples of 2-benzazepines fused with an additional furan ring are rare (for a recent synthesis, see [14]). Otherwise, **6a** could lose a H-atom to furnish a trivial *Mannich* product **8aa**. As shown in Table 1, the use of AlCl<sub>3</sub> or SbCl<sub>5</sub> as the *Lewis* acid was proved to be unsuccessful for triggering the expected reaction (Table 1, Entries 1 and 2). Interestingly, switching the *Lewis* acid to both Me<sub>2</sub>SiCl<sub>2</sub> and a mixture with AlCl<sub>3</sub> only led to **8aa** in comparable yields (Table 1, Entries 3 and 4). Finally, to our delight, by utilizing TiCl<sub>4</sub> as the *Lewis* acid, the reaction occurred smoothly to afford the desired product **7aa** as the only product in good yield (Table 1, Entry 5).

Having found the optimal *Lewis* acid for this reaction, it was necessary to optimize the reaction temperature. In our case, all reactions carried out from –78° to the ambient temperature led to the formation of **7aa** (Table 2), but cooling with ice bath (–5 to +5°) was adopted in terms of easy operation and high yield of the product.

Scheme 3

Table 1. Screening for the Appropriate Lewis Acid for the Synthesis of **7aa** and **8aa**<sup>a)</sup>

| Entry | Lewis acid  | Yield of <b>7aa</b> <sup>b)</sup> [%] | Yield of <b>8aa</b> <sup>b)</sup> [%] |
|-------|---|---------------------------------------|---------------------------------------|
| 1     | AlCl <sub>3</sub>   | 0                                     | 0                                     |
| 2     | SbCl <sub>5</sub>   | 0                                     | 0                                     |
| 3     | Me <sub>2</sub> SiCl <sub>2</sub>                             | 0                                     | 65                                    |
| 4     | Me <sub>2</sub> SiCl <sub>2</sub> / AlCl <sub>3</sub> (1 : 1) | 0                                     | 71                                    |
| 5     | TiCl <sub>4</sub>   | 81                                    | 0                                     |

<sup>a)</sup> Reaction conditions: **3a** (3 mmol), 2,3-dihydrofuran (**5a**; 3.9 mmol), Lewis acid (3 mmol), dry CH<sub>2</sub>Cl<sub>2</sub> (7 ml), –5 to +5°, 3 h. <sup>b)</sup> Yields of products isolated by silica gel column chromatography.

Table 2. Screening for the Optimal Reaction Temperature for the Synthesis of **7aa**<sup>a)</sup>

| Entry | T [°] | Yield <sup>b)</sup> [%] |
|-------|-------|-------------------------|
| 1     | –78   | 40                      |
| 2     | –40   | 81                      |
| 3     | –10   | 82                      |
| 4     | 20    | 71                      |

<sup>a)</sup> Reagents and reaction conditions were the same as provided in the footnote of Table 1. <sup>b)</sup> Yields obtained after chromatography.

Having established the optimum conditions, a number of different combinations of substrates were used to probe the scope and feasibility of this methodology for preparation of 2-benzazepine derivatives (Table 3).

Depending on the electronic nature of the starting aminoacetal **3** (Scheme 2), the reaction sequence afforded either the tricyclic 2-benzazepines **7** or a trivial uncyclized product **8**, or a mixture of 4-(hydroxyalkyl)-substituted 2,3-dihydro-2-benzazepines **9**. The formation of **9** may be accounted for by the acid-promoted ring opening of the

Table 3. Preparation of 2-Benzoazepine Derivatives by  $\text{TiCl}_4$ -promoted Tandem Mannich Reaction and Electrophilic Aromatic Substitution

| Entry             | <b>3</b>  | R                      | <b>5</b>  | <i>n</i> | Product    | Yield <sup>a</sup> ) [%] |
|-------------------|-----------|------------------------|-----------|----------|------------|--------------------------|
| 1                 | <b>3a</b> | 3,4-(MeO) <sub>2</sub> | <b>5a</b> | 1        | <b>7aa</b> | 82                       |
| 2                 | <b>3a</b> | 3,4-(MeO) <sub>2</sub> | <b>5b</b> | 2        | <b>7ab</b> | 68                       |
| 3                 | <b>3b</b> | 3-MeO                  | <b>5a</b> | 1        | <b>7ba</b> | 25                       |
|                   |           |                        |           |          | <b>9ba</b> | 31                       |
| 4                 | <b>3b</b> | 3-MeO                  | <b>5b</b> | 2        | <b>8bb</b> | 34                       |
|                   |           |                        |           |          | <b>9bb</b> | 35                       |
| 5                 | <b>3c</b> | 2,3-(MeO) <sub>2</sub> | <b>5a</b> | 1        | <b>9ca</b> | <sup>b</sup> )           |
| 6                 | <b>3c</b> | 2,3-(MeO) <sub>2</sub> | <b>5b</b> | 2        | <b>9cb</b> | <sup>b</sup> )           |
| 7 <sup>c</sup> )  | <b>3b</b> | 3-MeO                  | <b>5a</b> | 1        | <b>9ba</b> | 68                       |
| 8 <sup>c</sup> )  | <b>3b</b> | 3-MeO                  | <b>5b</b> | 2        | <b>9bb</b> | 72                       |
| 9 <sup>c</sup> )  | <b>3c</b> | 2,3-(MeO) <sub>2</sub> | <b>5a</b> | 1        | <b>9ca</b> | 56                       |
| 10 <sup>c</sup> ) | <b>3c</b> | 2,3-(MeO) <sub>2</sub> | <b>5b</b> | 2        | <b>9cb</b> | <sup>b</sup> )           |
| 11                | <b>3d</b> | H                      | <b>5a</b> | 1        | <b>8da</b> | 67                       |
| 12                | <b>3e</b> | 3-Me                   | <b>5a</b> | 1        | <b>8ea</b> | 70                       |

<sup>a</sup>) Yields of products isolated by silica gel column chromatography. <sup>b</sup>) Undetermined due to difficulty in purification. <sup>c</sup>) 2.0 Equiv. of  $\text{TiCl}_4$  were employed.

initially formed tricyclic products **7**. This illuminated that the attached additional furan or pyran ring of **7** did not tolerate acidic conditions. It was also shown that the amount of the employed  $\text{TiCl}_4$  affected the product distribution. As shown by the representative results in Table 3, the substrate **3a** bearing two electron-donating groups reacted with both 2,3-dihydrofuran (**5a**) and 3,4-dihydro-2H-pyran (**5b**) using 1 equiv. of  $\text{TiCl}_4$  as the Lewis acid to afford cleanly the tricyclic compounds **7aa** and **7ab** in 82 and 68% yield, respectively (Table 3, Entries 1 and 2). However, the use of **3b**, which holds only one MeO group, afforded a mixture of the tricyclic product **7ba** and the bicyclic product **9ba** in the case of **5a**, and **8bb** and **9bb** in the case of **5b** in essentially the same amount (Table 3, Entries 3 and 4). When using **3c** as substrate, the reaction gave intricate inseparable products (Table 3, Entries 5 and 6). Nevertheless, increasing the amount of the employed  $\text{TiCl}_4$  to two equivalents, the reaction using **3b** afforded the bicyclic fused heterocycles **9ba** and **9bb** in good yields without the detection of the uncyclized products **8** (Table 3, Entries 7 and 8). This suggested that excessive Lewis acid is beneficial to the concurrent aromatic electrophilic substitution. This was further verified by the reaction of **3c** bearing two electron-donating MeO groups at the C(2) and C(3) positions, which delivered again the bicyclic compounds **9ca** and **9cb**, respectively, though not in quite as good yields (Table 3, Entries 9 and 10). At this

point, it may be concluded that the presence of electron-donating groups on the benzene ring is helpful to the electrophilic aromatic substitution thus leading to the formation of the 2-benzazepine core. Accordingly, the reaction of **3d** as well as **3e** lacking a strong electron-donating MeO group gave only the trivial uncyclized products **9da** and **9ea**, respectively, in good yields (Table 3, Entries 11 and 12), and increasing the amount of  $\text{TiCl}_4$  to 2 equiv. did not give any cyclized products **7** or **9** (data not shown in Table 3).

The *trans* ring fusion between the attached furan or pyran ring, respectively, and the seven-membered ring in the tricyclic products **7aa**, **7ab**, and **7ba** has been tentatively assigned by inspection of the relatively large  $^1\text{H}, ^1\text{H}$ -coupling constant ( $> 9$  Hz) of  $\text{H}-\text{C}(3\text{a})/\text{C}(4\text{a})$  with  $\text{H}-\text{C}(10\text{b})/\text{C}(11\text{b})$ . Further proof to this stereochemical assignment came from the fact that no NOE for **7ab** could be observed between the angular H-atoms  $\text{H}-\text{C}(11\text{b})$  (apparent *d* at  $\delta(\text{H})$  4.28,  $J = 9.6$  Hz) and  $\text{H}-\text{C}(4\text{a})$  (*m* centered at  $\delta(\text{C})$  1.75).

Finally, we applied the developed novel method with the open-chain enol ether **5c** as a substrate (Table 4). The reaction with aminoacetals **3a–3c** proceeded smoothly under similar conditions. Notably, the isolated products were 2,3-dihydro-1*H*-benzazepines **10a–10c**. The formation of **10** may be explained by invoking the 4-ethoxy-2,3,4,5-tetrahydro-1*H*-2-benzazepines **11** (Figure) as intermediates *via* subsequent elimination of a molecule of EtOH.

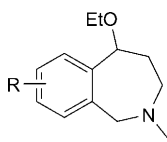
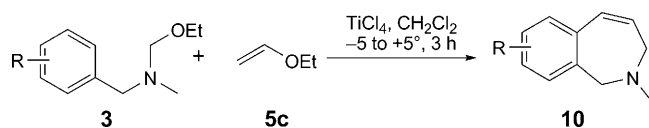
**11**

Figure. The proposed structure of intermediate **11**

Table 4. Preparation of 2,3-Dihydro-1*H*-2-benzazepines **10** Using **5c**<sup>a)</sup>



| Entry | <b>3</b>  | R                      | Product    | Yield <sup>a)</sup> [%] |
|-------|-----------|------------------------|------------|-------------------------|
| 1     | <b>3a</b> | 3,4-(MeO) <sub>2</sub> | <b>10a</b> | 71                      |
| 2     | <b>3b</b> | 3-MeO                  | <b>10b</b> | 67                      |
| 3     | <b>3c</b> | 2,3-(MeO) <sub>2</sub> | <b>10c</b> | 55                      |

<sup>a)</sup> 2.0 Equiv. of  $\text{TiCl}_4$  were employed. <sup>b)</sup> Yields of products isolated by silica gel column chromatography.

In conclusion, an efficient tandem synthesis of pharmaceutically important 2-benzazepines was developed. The protocol relies on  $\text{TiCl}_4$  mediated *Mannich*-type reaction of electron-rich benzyl iminium ions generated *in situ* from aminoacetals with alkenyl ethers such as 2,3-dihydrofuran, 3,4-dihydro-2*H*-pyran, or ethyl vinyl ether and concurrent *Friedel–Crafts*-type electrophilic substitution.

### Experimental Part

*General.*  $\text{CH}_2\text{Cl}_2$  was dried by refluxing over  $\text{CaH}_2$ , and other reagents were commercially available and used without further purification. Flash column chromatography (FC): silica gel 60 ( $\text{SiO}_2$ ; 230–400 mesh) with  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  mixture for elution. IR Spectra: in  $\text{CH}_2\text{Cl}_2$  or as KBr disc with a Nicolet-360 spectrophotometer; absorptions given in  $\text{cm}^{-1}$ .  $^1\text{H-NMR}$  Spectra: JEOL ECA 400; chemical shifts ( $\delta$ ) reported in ppm relative to TMS as internal standard; coupling constants ( $J$ ) in Hz. GC/MS: 30 m HP-5MS cap. column on a Hewlett-Packard 6890GC/5973MS system. HR-ESI-MS: SHIMADZU LCMS-IT-TOF.

*General Procedure for the Preparation of N-Benzyl-N-methylamines 2.* Gaseous  $\text{MeNH}_2$  was bubbled into a soln. of benzaldehyde **1** (0.05 mol) in benzene (70 ml) for 1 h under stirring, then the resulting mixture was heated to reflux, and the produced  $\text{H}_2\text{O}$  was removed by azeotropic distillation. Upon completion of the reaction (2 h), the mixture was evaporated *in vacuo*. The residue was dissolved in 70 ml of EtOH, and  $\text{NaBH}_4$  (2.27 g, 0.06 mol) was added in portions under stirring. After stirring at r.t. for 4 h, the solvent was removed *in vacuo*, the residue was dissolved in 200 ml of  $\text{CH}_2\text{Cl}_2$ , washed successively with  $\text{H}_2\text{O}$  and brine, dried ( $\text{Na}_2\text{SO}_4$ ), and filtered. The filtrate was concentrated *in vacuo* to give a yellow oil, which was subjected to vacuum distillation to afford the pure product **2**.

*General Procedure for Preparation of Amino Acetals 3.* Anh. *N*-benzyl-*N*-methylamine **2** (10 mmol), dry EtOH (2.3 ml, 4.0 mol-equiv.), and anh.  $\text{K}_2\text{CO}_3$  (1.51 g, 11 mmol, 1.1 mol-equiv.) were stirred at  $0^\circ$  for 15 min. Paraformaldehyde (0.33 g, 11 mmol, 1.1 mol-equiv.) was added in one portion, and the mixture was stirred at r.t. for 1 d. The reaction was monitored by TLC ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$  15 : 1). In case the reaction did not reach completion, an additional amount of paraformaldehyde (60 mg, 2 mmol) and 2 ml of anh.  $\text{Et}_2\text{O}$  were fed into the mixture, which then was stirred further overnight. The mixture was filtered, and the cake was washed several times with anh.  $\text{Et}_2\text{O}$ . The combined filtrate was concentrated *in vacuo* to give the product **3**, which could be used for the next step without purification. (Note: The products **3** are prone to decompose on exposure to moisture.)

*General Experimental Procedure for the Tandem Mannich and Aromatic Electrophilic Substitution Reactions.* To a soln. of **3** (3.0 mmol) in dry  $\text{CH}_2\text{Cl}_2$  (7 ml) was added the alkenyl ether **5** (3.9 mmol, 1.3 equiv.). After cooling to  $-5^\circ$ , anh.  $\text{TiCl}_4$  (as specified in Tables 3 and 4) was added dropwise over 10 min. After stirring for 3 h, the temp. was allowed gradually to warm to r.t., then the mixture was poured into a sat. aq.  $\text{NaHCO}_3$  soln. The org. phase was separated, the aq. phase was extracted with  $\text{CH}_2\text{Cl}_2$  or AcOEt ( $3 \times 60$  ml). The org. phases were combined, dried ( $\text{Na}_2\text{SO}_4$ ), and concentrated under reduced pressure. The residue was purified by CC ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$  1 : 1) to give the products **7**, **8**, **9**, or **10**.

*3,3a,4,5,6,10b-Hexahydro-8,9-dimethoxy-5-methyl-2H-furo[3,2-d][2]benzazepine (7aa).* Eluent  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  60 : 1. Colorless oil. Yield: 82%. IR (KBr): 3400, 2926, 1607, 1514, 1463, 1276, 1114, 1029.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.03 (s, 1 H); 6.57 (s, 1 H); 5.13 (d,  $J = 9.2$ , 1 H); 3.99–3.95 (m, 1 H); 3.81 (s, 3 H); 3.79 (s, 3 H); 3.73 (d,  $J = 13.3$ , 1 H); 3.70–3.63 (m, 1 H); 3.37 (d,  $J = 13.3$ , 1 H); 2.70–2.62 (m, 1 H); 2.54 (dd,  $J = 2.5, 12.6$ , 1 H); 2.22 (s, 3 H); 2.02–1.96 (m, 1 H); 1.90 (t,  $J = 12.1$ , 1 H); 1.33–1.24 (m, 1 H).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 148.4; 147.3; 131.9; 121.2; 113.1; 108.5; 79.5; 67.0; 56.1; 55.9; 55.8; 55.6; 43.2; 39.0; 31.9. GC/MS: 263 ( $M^+$ ). HR-ESI-MS: 264.1662 ( $[M + H]^+$ ,  $\text{C}_{15}\text{H}_{22}\text{NO}_3^+$ ; calc. 264.1600).

*2,3,4,4a,5,6,7,11b-Octahydro-9,10-dimethoxy-6-methylpyrano[3,2-d][2]benzazepine (7ab).* Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  60 : 1. Colorless solid. Yield: 68%. IR (KBr): 3431, 2928, 2837, 1604, 1514, 1444, 1270, 1213, 1114, 1027, 821.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.31 (s, 1 H); 6.62 (s, 1 H); 4.28 (d,  $J = 9.6$ , 1 H); 4.20–4.14 (m, 1 H); 3.96 (s, 3 H); 3.85 (s, 3 H); 3.82–3.80 (m, 1 H); 3.53 (d,  $J = 14.2$ , 1 H); 3.45–3.39 (m, 1 H); 2.85 (d,  $J = 14.2$ , 1 H); 2.72–2.66 (m, 1 H); 2.22 (s, 3 H); 1.78–1.67 (m, 3 H); 1.62–1.59 (m, 1 H); 1.21–1.17 (m, 1 H).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 147.8; 146.7; 135.2; 127.5; 113.3; 109.0; 79.8; 67.5; 65.9; 61.4; 56.0 (2 MeO); 42.8; 36.2; 29.2; 25.5. HR-ESI-MS: 278.1744 ( $[M + H]^+$ ,  $\text{C}_{16}\text{H}_{24}\text{NO}_3^+$ ; calc. 278.1756).

*2,3,3a,4,5,6-Hexahydro-8-methoxy-5-methyl-2H-furo[3,2-d][2]benzazepine (7ba).* Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  50 : 1. Colorless oil. Yield: 25%. IR (KBr): 3455, 2943, 2617, 1613, 1505, 1481, 1254, 1069.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.33 (d,  $J = 8.7$ , 1 H); 6.77 (dd,  $J = 2.3, 8.7$ , 1 H); 6.60 (d,  $J = 2.3$ , 1 H); 5.10 (d,  $J = 9.2$ , 1 H); 3.96–3.92 (m, 1 H); 3.83 (d,  $J = 13.5$ , 1 H); 3.70 (s, 3 H); 3.67–3.61 (m, 1 H); 3.46 (d,  $J = 13.5$ , 1 H); 2.74–2.69 (m, 1 H); 2.60 (dd,  $J = 2.3, 12.4$ , 1 H); 2.26 (s, 3 H); 2.03–1.96 (m, 1 H); 1.91 (t,

$J = 12.4, 1 \text{ H}$ ); 1.35–1.29 ( $m, 1 \text{ H}$ ).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 158.7; 131.5; 129.5; 116.0; 115.9; 112.5; 79.5; 67.0; 56.2; 55.5; 55.3; 43.0; 39.2; 31.8. HR-ESI-MS: 234.1487 ( $[M+H]^+$ ,  $\text{C}_{14}\text{H}_{20}\text{NO}_2^+$ ; calc. 234.1494).

*1-(4,5-Dihydrofuran-3-yl)-N-(3,4-dimethoxybenzyl)-N-methylmethanamine (8aa)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  80:1. Colorless oil. Yield: 71%.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 6.89 ( $s, 1 \text{ H}$ ); 6.80 ( $s, 2 \text{ H}$ ); 6.24 ( $s, 1 \text{ H}$ ); 4.35 ( $t, J = 9.6, 2 \text{ H}$ ); 3.88 ( $s, 3 \text{ H}$ ); 3.84 ( $s, 3 \text{ H}$ ); 3.41 ( $s, 2 \text{ H}$ ); 3.00 ( $s, 2 \text{ H}$ ); 2.64 ( $t, J = 9.6, 2 \text{ H}$ ); 2.17 ( $s, 3 \text{ H}$ ).

*1-(3,4-Dihydro-2H-pyran-5-yl)-N-(3-methoxybenzyl)-N-methylmethanamine (8bb)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  80:1. Colorless oil. Yield: 34%. IR (KBr): 2940, 2782, 1667, 1601, 1488, 1264, 1155, 1051, 877.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 6.99 ( $d, J = 8.2, 1 \text{ H}$ ); 6.67 ( $dd, J = 8.2, 2.3, 1 \text{ H}$ ); 6.59 ( $d, J = 2.3, 1 \text{ H}$ ); 6.24 ( $s, 1 \text{ H}$ ); 3.73 ( $s, 3 \text{ H}$ ); 3.68 ( $s, 2 \text{ H}$ ); 3.56 ( $t, J = 6.4, 2 \text{ H}$ ); 3.28 ( $s, 2 \text{ H}$ ); 2.31 ( $s, 3 \text{ H}$ ); 2.13 ( $t, J = 7.6, 2 \text{ H}$ ); 1.71–1.67 ( $m, 2 \text{ H}$ ).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 158.1; 139.1; 138.3; 131.3; 129.6; 126.2; 114.8; 112.2; 61.9; 61.3; 60.5; 55.3; 42.6; 34.6; 32.0.

*2-(2,3-Dihydro-8-methoxy-2-methyl-1H-2-benzazepin-4-yl)ethanol (9ba)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  20:1. Yellow oil. Yield: 68%. IR (KBr): 3394, 2939, 1609, 1504, 1466, 1254, 1043.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.09 ( $d, J = 8.7, 1 \text{ H}$ ); 6.78 ( $dd, J = 2.7, 8.7, 1 \text{ H}$ ); 6.68 ( $d, J = 2.7, 1 \text{ H}$ ); 6.41 ( $s, 1 \text{ H}$ ); 3.79 ( $s, 3 \text{ H}$ ); 3.76 ( $t, J = 6.0, 2 \text{ H}$ ); 3.70 ( $s, 2 \text{ H}$ ); 3.16 ( $s, 2 \text{ H}$ ); 2.41 ( $t, J = 6.0, 2 \text{ H}$ ); 2.40 ( $s, 3 \text{ H}$ ).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 158.4; 137.9; 137.1; 130.9; 130.0; 128.8; 115.2; 112.4; 62.5; 60.1; 60.0; 55.4; 43.1; 41.3. HR-ESI-MS: 234.1490 ( $[M+H]^+$ ,  $\text{C}_{14}\text{H}_{20}\text{NO}_2^+$ ; calc. 234.1494).

*3-(2,3-Dihydro-8-methoxy-2-methyl-1H-2-benzazepin-4-yl)propan-1-ol (9bb)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  25:1. Yellow oil. Yield: 72%. IR (KBr): 3411, 2934, 1608, 1501, 1278, 1248, 1034.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.05 ( $d, J = 8.7, 1 \text{ H}$ ); 6.74 ( $dd, J = 1.8, 8.7, 1 \text{ H}$ ); 6.65 ( $d, J = 1.8, 1 \text{ H}$ ); 6.30 ( $s, 1 \text{ H}$ ); 3.78 ( $s, 3 \text{ H}$ ); 3.75 ( $s, 2 \text{ H}$ ); 3.64 ( $t, J = 6.4, 2 \text{ H}$ ); 3.34 ( $s, 2 \text{ H}$ ); 2.37 ( $s, 3 \text{ H}$ ); 2.20 ( $t, J = 7.8, 2 \text{ H}$ ); 1.79–1.73 ( $m, 2 \text{ H}$ ).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 158.1; 139.1; 138.3; 131.3; 129.6; 126.2; 114.8; 112.2; 61.9; 60.5; 55.3; 42.6; 34.6; 32.0.

*2-(2,3-Dihydro-8,9-dimethoxy-2-methyl-1H-2-benzazepin-4-yl)ethanol (9ca)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  20:1. Yellow oil. Yield: 56%.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 6.89 ( $d, J = 8.7, 1 \text{ H}$ ); 6.85 ( $d, J = 8.7, 1 \text{ H}$ ); 6.56 ( $s, 1 \text{ H}$ ); 3.93 ( $s, 2 \text{ H}$ ); 3.84 ( $s, 3 \text{ H}$ ); 3.80 ( $t, J = 7.9, 2 \text{ H}$ ); 3.78 ( $s, 3 \text{ H}$ ); 3.14 ( $s, 2 \text{ H}$ ); 2.55 ( $s, 3 \text{ H}$ ); 2.46 ( $t, J = 7.9, 2 \text{ H}$ ).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 151.6; 147.5; 136.3; 131.5; 131.4; 126.3; 124.8; 112.2; 61.4; 57.3; 55.9; 50.2; 46.0; 42.6; 40.8.

*N-Benzyl-1-(4,5-dihydrofuran-3-yl)-N-methylmethanamine (8da)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  80:1. Colorless oil. Yield: 67%.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.29–7.21 ( $m, 5 \text{ H}$ ); 6.22 ( $s, 1 \text{ H}$ ); 4.32 ( $t, J = 9.4, 2 \text{ H}$ ); 3.44 ( $s, 2 \text{ H}$ ); 3.00 ( $s, 2 \text{ H}$ ); 2.63 ( $t, J = 9.4, 2 \text{ H}$ ); 2.14 ( $s, 3 \text{ H}$ ).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 143.1; 139.5; 129.0; 128.3; 127.0; 112.7; 70.4; 61.4; 53.7; 42.4; 31.6. GC/EI-MS: 203.1 ( $M^+$ ).

*1-(4,5-Dihydrofuran-3-yl)-N-methyl-N-(3-methylbenzyl)ethanamine (8ea)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  80:1. Colorless oil. Yield: 70%.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.20–7.05 ( $m, 4 \text{ H}$ ); 6.25 ( $s, 1 \text{ H}$ ); 4.36 ( $t, J = 9.6, 2 \text{ H}$ ); 3.43 ( $s, 2 \text{ H}$ ); 3.03 ( $s, 2 \text{ H}$ ); 2.69–2.64 ( $m, 2 \text{ H}$ ); 2.35 ( $s, 3 \text{ H}$ ); 2.17 ( $s, 3 \text{ H}$ ).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 143.2; 139.3; 137.9; 129.7; 128.2; 127.7; 126.0; 112.7; 70.5; 61.4; 53.8; 42.4; 31.6; 21.5. GC/EI-MS: 217.2 ( $M^+$ ).

*2,3-Dihydro-7,8-dimethoxy-2-methyl-1H-2-benzazepine (10a)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  60:1. Yellow oil. Yield: 71%. IR (KBr): 3393, 2968, 1604, 1520, 1466, 1353, 1274, 1116.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 6.70 ( $s, 1 \text{ H}$ ); 6.64 ( $s, 1 \text{ H}$ ); 6.37 ( $d, J = 12.4, 1 \text{ H}$ ); 5.70 ( $dt, J = 3.7, 12.4, 1 \text{ H}$ ); 3.88 ( $s, 3 \text{ H}$ ); 3.86 ( $s, 3 \text{ H}$ ); 3.78 ( $s, 2 \text{ H}$ ); 3.51 ( $s, 2 \text{ H}$ ); 2.41 ( $s, 3 \text{ H}$ ).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 147.7; 147.6; 131.1; 129.4; 129.1; 129.0; 113.8; 112.5; 61.1; 59.8; 56.1 (2 MeO); 43.3. HR-ESI-MS: 220.1312 ( $[M+H]^+$ ,  $\text{C}_{13}\text{H}_{18}\text{NO}_2^+$ ; calc. 220.1337).

*2,3-Dihydro-8-methoxy-2-methyl-1H-2-benzazepine (10b)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  70:1. Yellow oil. Yield: 67%. IR (KBr): 3402, 2958, 2524, 1608, 1504, 1466, 1258, 1031, 874.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 7.09 ( $d, J = 8.2, 1 \text{ H}$ ); 6.72 ( $dd, J = 2.3, 8.2, 1 \text{ H}$ ); 6.65 ( $d, J = 2.3, 1 \text{ H}$ ); 6.36 ( $d, J = 10.5, 1 \text{ H}$ ); 5.64–5.59 ( $m, 1 \text{ H}$ ); 3.82 ( $s, 2 \text{ H}$ ); 3.79 ( $s, 3 \text{ H}$ ); 3.52 ( $s, 2 \text{ H}$ ); 2.39 ( $s, 3 \text{ H}$ ).  $^{13}\text{C-NMR}$  (100 MHz,  $\text{CDCl}_3$ ): 158.6; 140.0; 132.0; 129.2; 129.1; 128.2; 114.8; 111.8; 61.7; 60.2; 55.3; 43.3. HR-ESI-MS: 190.1242 ( $[M+H]^+$ ,  $\text{C}_{12}\text{H}_{16}\text{NO}^+$ ; calc. 190.1232).

*2,3-Dihydro-8,9-dimethoxy-2-methyl-1H-2-benzazepine (10c)*. Eluent:  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  70:1. Yellow oil. Yield: 55%.  $^1\text{H-NMR}$  (400 MHz,  $\text{CDCl}_3$ ): 6.90 ( $d, J = 8.2, 1 \text{ H}$ ); 6.75 ( $d, J = 8.2, 1 \text{ H}$ ); 6.39 ( $d,$

$J = 11.9, 1 \text{ H}$ ;  $5.68\text{--}5.63 \text{ (}m, 1 \text{ H)}$ ;  $3.90 \text{ (}s, 2 \text{ H)}$ ;  $3.85 \text{ (}s, 3 \text{ H)}$ ;  $3.78 \text{ (}s, 3 \text{ H)}$ ;  $2.19 \text{ (}s, 2 \text{ H)}$ ;  $2.48 \text{ (}s, 3 \text{ H)}$ .  
GC/EI-MS: 219.2 ( $M^+$ ).

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